This article was downloaded by:

On: 26 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



## Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

# INCREASED RNA AFFINITY OF HNA ANALOGUES BY INTRODUCING ALKOXY SUBSTITUENTS AT THE C-1 OR C-3 POSITION

A. Van Aerschot<sup>a</sup>; M. Meldgaard<sup>a</sup>; F. Volders<sup>a</sup>; G. Schepers<sup>a</sup>; J. Rozenski<sup>a</sup>; P. Herdewijn<sup>a</sup> Rega Institute for Medical Research, Katholieke Universiteit Leuven, Leuven, Belgium

Online publication date: 31 March 2001

To cite this Article Van Aerschot, A. , Meldgaard, M. , Volders, F. , Schepers, G. , Rozenski, J. and Herdewijn, P.(2001) INCREASED RNA AFFINITY OF HNA ANALOGUES BY INTRODUCING ALKOXY SUBSTITUENTS AT THE C-1 OR C-3 POSITION', Nucleosides, Nucleotides and Nucleic Acids, 20: 4, 781-784

To link to this Article: DOI: 10.1081/NCN-100002429 URL: http://dx.doi.org/10.1081/NCN-100002429

### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

# INCREASED RNA AFFINITY OF HNA ANALOGUES BY INTRODUCING ALKOXY SUBSTITUENTS AT THE C-1 OR C-3 POSITION

A. Van Aerschot,\* M. Meldgaard, F. Volders, G. Schepers, J. Rozenski, and P. Herdewijn

Rega Institute for Medical Research, Katholieke Universiteit Leuven, Minderbroedersstraat 10, 3000 Leuven, Belgium

### **ABSTRACT**

1,5-Anhydrohexitol nucleoside congeners with alkoxy substituents, were prepared, resulting in a further improvement of their RNA affinity and antisense potential.

The strong hybridizing potential of anhydrohexitol nucleic acids (HNA) by virtue of its pre-organisation by now is well documented (1–3), and some interesting biological antisense effects have been reported (inhibition of Ha-ras and ICAM-1 (4), and antimalarial activity (5)). To further augment the affinity for target RNA structures, two different roads can be explored by looking for analogues which either increase the conformational preorganisation of the monomeric structures, or which alternatively augment the hydration potential. In addition, cost and ease of synthesis need to be considered. Consequently, 3'-O-methylated altrohexitol analogues and 1'-O-methylglycosidic analogues were envisaged and incorporated into HNA sequences.

Synthesis of the 3'-O-methylated analogue followed the route previously described for preparation of the altrohexitol monomers (ANA) (6). Chemoselective methylation without temporary protection of the nucleobase gave the methylated nucleoside. Further modification yielded the desired phosphitylated building block  $1^*$ .

<sup>\*</sup>Corresponding author.



2 VAN AERSCHOT ET AL.

The 1'-O-methylglycosidic analogues were obtained starting from ubiquitous methyl glucopyranoside. Opening of the 2,3-allo-epoxide (7,8), was followed by deoxygenation of the 3-position. Removal of the benzylidene position is less straightforward because of the glycosidic linkage, but can be accomplished alternatively via hydrogenation in almost quantitative yield. Further modification yielded the desired phosphitylated building block 2\*.

All oligos were assembled on a propanediol containing universal support, obviating the need of modified supports (9). With a coupling time of 3 minutes, coupling yields were consistently over 95% and higher. Electrospray ionization mass spectrometry (ESI-MS) in negative mode indicated all monoisotopic masses to be consistently within 0.5 Da of the calculated masses.

 $\begin{array}{lll} \textbf{Scheme 1.} & \text{a) } 3.2 \text{ eq. uracil, } 3 \text{ eq. NaH, DMF } 120^{\circ}\text{C } 24\text{h } (86\%); \text{ b) } 5 \text{ eq. NaH, } 3 \text{ eq. CH}_{3}\text{I, THF, } 7\text{h } 0^{\circ}\text{C } (38\%); \text{ c) } 90\% \text{ TFA } (74\%); \text{ d) } \text{MMTrCl, pyridine } (89\%); \text{ e) DIEA, } \text{CH}_{2}\text{Cl}_{2}, \text{ (iPr)}_{2}\text{N(CN)PCl } (90\%); \text{ f) } 3 \text{ eq. thymine, } 2.8 \text{ eq. NaH, DMF, } 96\text{ h, } 120^{\circ}\text{C } (71\%); \text{ g) } 2 \text{ eq. CSCl}_{2}, 7 \text{ eq. DMAP, } \text{CH}_{2}\text{Cl}_{2} \text{ at } -40^{\circ}\text{C } \text{ followed by } 4 \text{ eq. } 2,4\text{-Cl}_{2}\text{C}_{6}\text{H}_{3}\text{OH } \text{ at } \text{RT for } 1\text{ h } (85\%); \text{ h) } 1.5 \text{ eq. Bu}_{3}\text{SnH, AIBN, toluene } 80^{\circ}\text{C } (85\%); \text{ i) } 10\% \text{ TFA-MeOH } 3\text{ h } (55\%); \text{ alternatively } \text{H}_{2}, \text{Pd/C in MeOH-HOAc } 98:2 \text{ for } 18\text{ h } (90\%); \text{ j) } \text{DMTrCl, pyridine } (85\%); \text{ k) } \text{DIEA, } \text{CH}_{2}\text{Cl}_{2}, \text{ (iPr)}_{2}\text{N(CN)PCl } (67\%). \\ \end{array}$ 

When hybridised with complementary HNA, the introduction of the 3'-O-methylated uridine nucleoside analogue (1) into a HNA strand results in an increased thermal stability of the duplex compared to the unmodified HNA:HNA duplex ( $\Delta T_m = +0.6^{\circ}$ C/modification) (entry **A** and **B**, Table 1). However, this increase is less pronounced than the increase in thermal stability obtained by modifying the nucleobase with a methyl substituent in the 5-position ( $\Delta T_m = +1.1^{\circ}$ C/



#### INCREASED RNA AFFINITY OF HNA ANALOGUES

**Table 1.** Hybridisation Data for Hexameric Hexitol Sequences  $(6' \rightarrow 4')$  with Incorporation of 3 Methylated Building Blocks 1 or 2

	Sequence	HNA Complement	ANA Complement	RNA Complement	DNA Complement
A	1 C 1 CC 1 (HNA)	52.4 (64)	58.8 (71)	31 (42)	no T <sub>m</sub>
B	UCU CCU (HNA)	50.7 (61.2)	55#	30.5 (40)	
C	UCU CCU (ANA)	54	61.8 (71.2)	38.4 (47.6)	no T <sub>m</sub>
D	TCT CCT (HNA)	54	60.6	39 (48)	$\begin{array}{c} \text{no } T_m \\ \text{no } T_m \end{array}$
E	2 C 2 CC 2 (HNA)	56.7	62.7	39.9 (49.5)	

 $T_m$ 's in a buffer of 0.1 M NaCl and 20 mM phosphate, pH 7.4 with a duplex concentration of 4  $\mu$ M. Numbers in brackets are T<sub>m</sub>'s in a high salt buffer (1.0 M NaCl). 1 denotes a 3'-O-methylated ANA monomer, 2 denotes a 1'-O-methylated HNA monomer.

# lit data incorrect (in ref. 12, the duplex was HNA-HNA, instead of HNA-ANA.

modification) (compare entry A and D). Hybridising the modified ONs with complementary RNA corroborates this pattern of thermal stabilisation of the duplexes. As expected, none of the hexitol based ONs hybridised with complementary DNA. Clearly, for this series the results indicate that a 5-methyl is more important than a 3'-O-methyl and that a methylation of the 3'-hydroxyl group in ANA is destabilising when pairing to an RNA sequence is envisaged.

For the hexitol analogue 2 comparison is more straightforward, and the thermal stabilisation is of the same order as for 1 when compared to HNA, as well in its pairing with hexitol oligonucleotides as with RNA sequences. Therefore, introduction of 2 seems to be slightly more favorable over addition of a HNA monomer.

Incorporation of respectively 1 or 2 into octameric (GCGUAGCG) HNA sequences and hybridisation with RNA, yielded an analogous picture with a small increase for introduction of 1 and a solid increase of 1.6°C in T<sub>m</sub> for introduction of 2. However, thermal unwinding of a self-complementary HNA duplex gave another pattern, where incorporation of 1 resulted in considerable stabilisation of the duplex  $(\Delta T_m = +3^{\circ}C/modification)$  exceeding the stabilisation obtained for substitution of uracil for thymine ( $\Delta T_m = +2.4$  °C/modification) (Table 2).

Table 2. Thermal Stability of Self Complementary HNA Sequences Containing 3'-O-methyl or 1'-O-methyl Modifications (1 or 2)

Sequence	$T_m/^{\circ}C$	$\Delta T_m/^{\circ}C^a$
GUGU ACAC	65.0	ref.
G 1 G 1 ACAC	76.7	+3
GTGT ACAC	74.5	+2.4/ref.
G 2 G 2 ACAC	76.9	+0.6

T<sub>m</sub>'s obtained in 0.1 M NaCl, 20 mM phosphate, pH 7.4 with an oligo concentration of 8  $\mu$ M (4  $\mu$ M of duplex).



 $<sup>^{</sup>a}\Delta T_{m}$ /modification.

784 VAN AERSCHOT ET AL.

Generally, the ONs containing the 3'-O-methyl derivative (1) showed a small increase in thermal stability towards complementary sequences as compared to HNA, except in the case of a self-complementary sequence for which an increase in thermal stability of 3°C per modification was observed. Compared to ANA, however, the 3'-O-methylation caused a small decrease in thermal stability of duplexes between a modified ON and a complementary target, especially when targeting RNA. The methyl glycosidic analogues 2, however, seem to be endowed with higher affinity for RNA in comparison with well-known HNA, while at the same time having economically more favorable monomers. These compounds therefore could have strong potential for antisense purposes and need to be evaluated further.

In addition, it proved possible to incorporate the new modified monomers into RNA without compromising their affinity for their respective RNA target.

#### ACKNOWLEDGMENT

This work was supported by a grant from the K.U. Leuven (GOA 97/11).

### REFERENCES

- 1. Hendrix, C.; Rosemeyer, H.; Verheggen, I.; Seela, F.; Van Aerschot, A.; Herdewijn, P. *Chemistry: European J.*, **3**, 110–120 (1997).
- 2. Hendrix, C.; Rosemeyer, H.; De Bouvere, B.; Van Aerschot, A.; Seela, F.; Herdewijn, P. *Chemistry: European J.*, **3**, 1513–1520 (1997).
- 3. Boudou, V.; Kerremans, L.; De Bouvere, B.; Schepers, G.; Busson, R.; Van Aerschot, A.; Herdewijn, P. *Nucleic Acids Res.*, **27**, 1450–1456 (1999).
- 4. Vandermeeren, M.; Preveral, S.; Janssens, S.; Geysen, J.; Saison-Behmoaras, E.; Van Aerschot, A.; Herdewijn, P. *Biochem Pharmacol.*, **59**, 655–663 (2000).
- 5. Flores, Maria V.C.; Atkins, D.; Stewart, T.S.; Van Aerschot, A.; Herdewijn, P. *Parasitol. Res. Series*, **85**, 864–866, (1999).
- 6. Allart, B.; Busson, R.; Rozenski, J.; Van Aerschot, A.; Herdewijn, P. *Tetrahedron*, **55**, 6527–6546 (1999).
- 7. Richtmeyer, N.K.; Hudson, C.S. C.S.; J. Am. Chem. Soc., 63, 1727–1731 (1941).
- 8. Rosenfeld, D.A.; Richtmeyer, N.K.; Hudson, C.S. J. Am. Chem. Soc., **70**, 2201–2206 (1948).
- 9. Van Aerschot, A.; Saison-Behmoaras, E.; Rozenski, J.; Hendrix, C.; Schepers, G.; Verhoeven, G.; Herdewijn, P. Bull. Soc. Chim. Belges, **104**, 717–720 (1995).



# **Request Permission or Order Reprints Instantly!**

Interested in copying and sharing this article? In most cases, U.S. Copyright Law requires that you get permission from the article's rightsholder before using copyrighted content.

All information and materials found in this article, including but not limited to text, trademarks, patents, logos, graphics and images (the "Materials"), are the copyrighted works and other forms of intellectual property of Marcel Dekker, Inc., or its licensors. All rights not expressly granted are reserved.

Get permission to lawfully reproduce and distribute the Materials or order reprints quickly and painlessly. Simply click on the "Request Permission/Reprints Here" link below and follow the instructions. Visit the U.S. Copyright Office for information on Fair Use limitations of U.S. copyright law. Please refer to The Association of American Publishers' (AAP) website for guidelines on Fair Use in the Classroom.

The Materials are for your personal use only and cannot be reformatted, reposted, resold or distributed by electronic means or otherwise without permission from Marcel Dekker, Inc. Marcel Dekker, Inc. grants you the limited right to display the Materials only on your personal computer or personal wireless device, and to copy and download single copies of such Materials provided that any copyright, trademark or other notice appearing on such Materials is also retained by, displayed, copied or downloaded as part of the Materials and is not removed or obscured, and provided you do not edit, modify, alter or enhance the Materials. Please refer to our Website User Agreement for more details.

# **Order now!**

Reprints of this article can also be ordered at http://www.dekker.com/servlet/product/DOI/101081NCN100002429